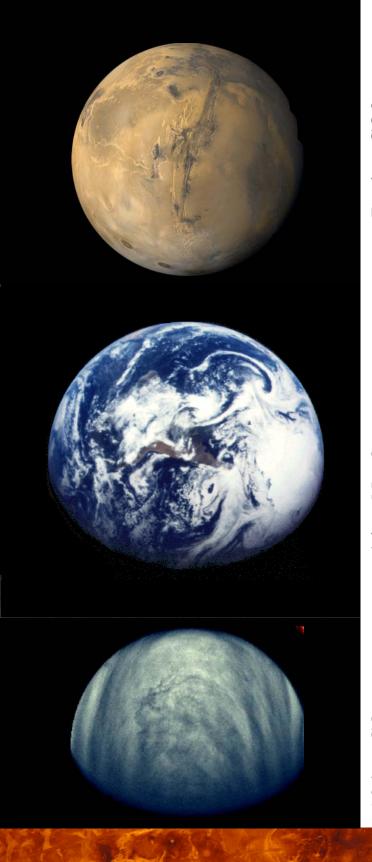
Neutral Mass Spectrometry for Venus Atmosphere and Surface

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Why such divergent evolution in terrestrial planets?



90 bar CO₂
730 K
H₂SO₄ clouds
100,000 x drier
than Earth
D/H 160 x Earth
(Venus once wet?)
Thermochemistry
below clouds

1 bar N₂, O₂
300 K in San Francisco
Receives ½ the solar
radiation of Venus
H₂O clouds
Oceans, Life

7 mbar CO2
~210 K
H₂O and CO₂
ice clouds
D/H 5 x Earth
Photochemistry

at surface

How unique is our solar system?

Motivation for improved mass spectrometer measurements at Venus

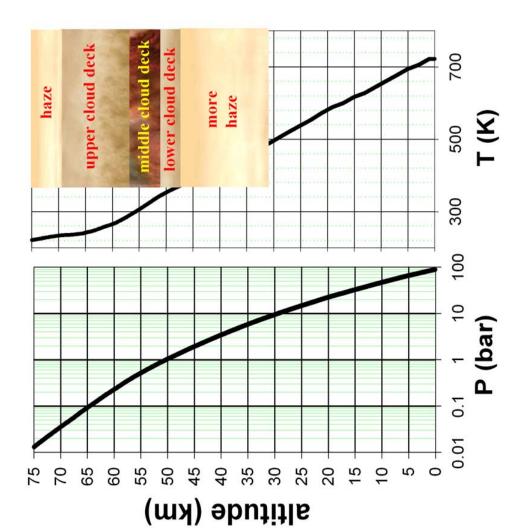
to address fundamental issues of terrestrial planetary formation and evolution

The assignment

- to make precise (better than 1 %) measurements of isotope ratios and accurate (5-10%) measurements of abundances of noble gas
- to obtain vertical profiles of trace chemically active gases from above the clouds all the way down to the surface

The challenge for Venus probe mass spectrometry

- 4 orders of magnitude pressure differential on track from above clouds to surface
- trace species measured to parts per billion
- 9 orders of magnitude difference between atmospheric pressure at surface and ion source pressure in mass spectrometer
- 500 degree temperature gradient from atmosphere above clouds to surface
- cloud droplets and aerosols that can clog mass spectrometer inlet systems and mask real vertical variations due to their condensation on surfaces
- a fast ride to the surface with an entry probe



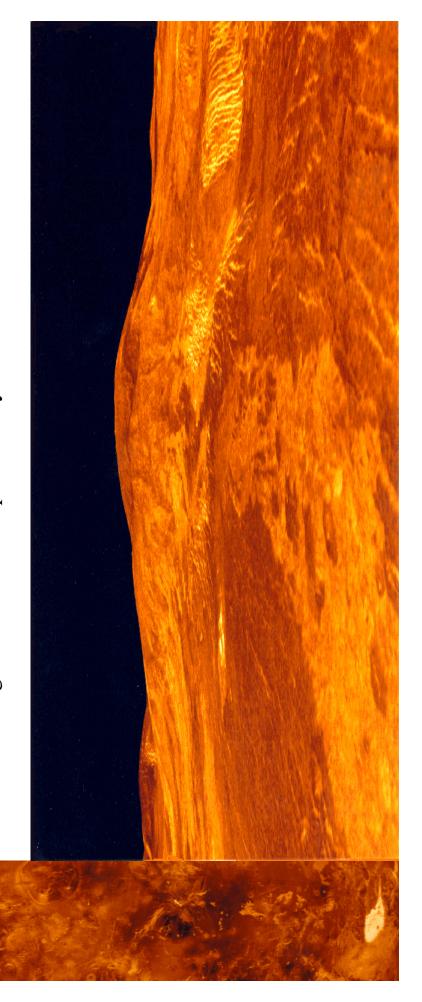


Near term Venus science goals for chemical and isotopic measurements

Where have the Venus missions, to date, left us with respect to these goals?

- noble gas elemental and isotopic composition
- cloud chemistry
 - surface science

The challenge of Venus mass spectrometry and future directions



Science goals - atmosphere & surface chemical & isotope measurements

Space Studies Board SSE Strategy – July 2002

- The first billion years of solar system history
- 1. What processes marked the initial stages of planet and satellite formation?
- 2. How long did it take Jupiter to form and how did the formation of the gas and ice giants differ?
- 3. What was the rate of decrease of impacts by comets, asteroids, and other objects and how did it affect the emergence of life?

Volatiles and organics: the stuff of life

- 1. What is the history of volatile material, especially water, in our solar system?
- What is the nature and history of organic material in our solar system?
- 3. What planetary processes affect the evolution of volatile on planets?

The origin and evolution of habitable worlds

- 1. Where are zones in our solar system where like can exist and what are the processes for producing and sustaining habitable planets?
- 2. Does (or did) life exist beyond the Earth?
- 3. Why did Mercury, Venus, Earth, and Mars diverge so much in their evolution?
- 4. What hazards do solar system objects present to Earth?

How planets work

- 1. How do the processes that shape planets today operate and interact?
- What does our solar system tell us about other solar systems?

Decadal Study Recommendations for Venus

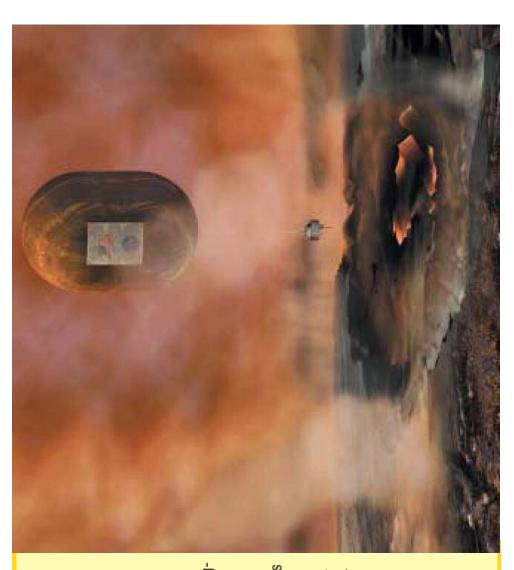
Profile Venus In Situ Explorer

Mission Type: Lander

Cost Class: Medium

Priority Measurements:

- Determine elemental and mineralogical surface compositions.
- atmospheres, especially trace gases and their isotopes. Measure the composition of the
- ments of noble gases and light sta-ble isotopes. Undertake high-precision measure-
- atmosphere-surface interaction. Assess processes and rates of
- gases in inner-planet atmospheres. Search for evidence of volcanic



Decadal Study Themes and Science Questions for Terrestrial Planets

Suiding Themes Addressed

Important Planetary Science Questions Addressed

Volatiles and Organics
The Stuff of Life

What global mechanisms affect the evolution of volatiles on planetary bodies?

What is the history of water on the inner planets? How did the atmospheres of the inner planets evolve?

> The Origin and Evolution of Habitable Worlds

Why have the terrestrial planets differed so dramatically in their evolution?

What kinds of minerals are the inner planets made of, and does this vary depending on a planet's distance from the Sun?

Processes
How Planetary
Systems Work

How do the processes that shape the contemporary character of planetary bodies operate and interact?

What processes stabilize climate?

How do planets' varied geological histories enable predictions of volacanic and tectonic activity?

Science measurement objectives of VISE are as follows:

- Determine the composition of Venus' atmosphere, including trace gas species and light stable isotopes
- Accurately measure noble-gas isotopic abundance in the atmosphere
- Provide descent, surface, and ascent meteorological data
- Measure zonal cloud-level winds over several Earth days
- Obtain near-IR descent images of the surface from 10-km altitude to the surface
- Accurately measure elemental abundances & mineralogy of a core from the surface
- Evaluate the texture of surface materials to constrain weathering environment.

Motivation for noble gas measurements at Venus

constrain models of atmospheric formation and evolution Noble gas elemental ratios and isotopic fractionation



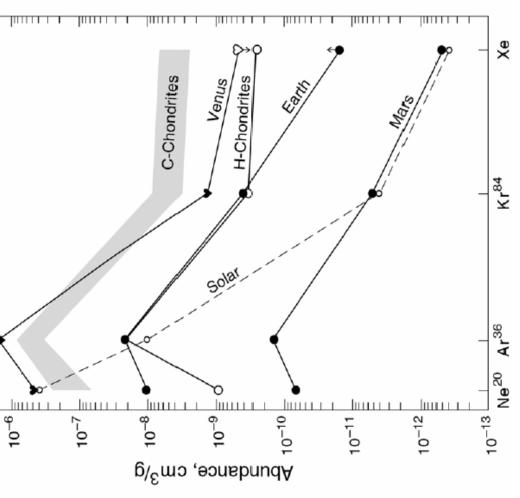
Noble gas elemental ratios

10⁻⁵巨

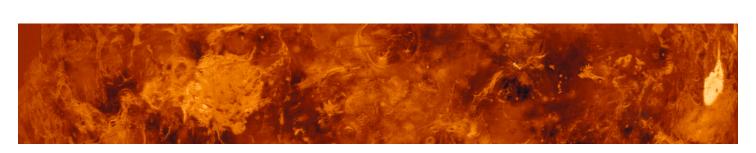
Inner planet noble gas elemental abundances do not match those of the sun or various types of chondrites.

The ³⁶Ar/⁸⁴Kr ratio at Venus may be much more solar like than Earth or Mars.

However - great uncertainty in Kr and Xe elemental abundances



From Owen and Bar-Num, Orig. of Life and the Evolution of the Biosphere, 31, 435, 2001.



Xenon Isotopic Composition



The Martian values are established from SNC meteorite analysis.

200

MEASURED COMPOSITIONS

8^MXe (‰)

300

XENON ISOTOPES

Mars (EETA 79001)

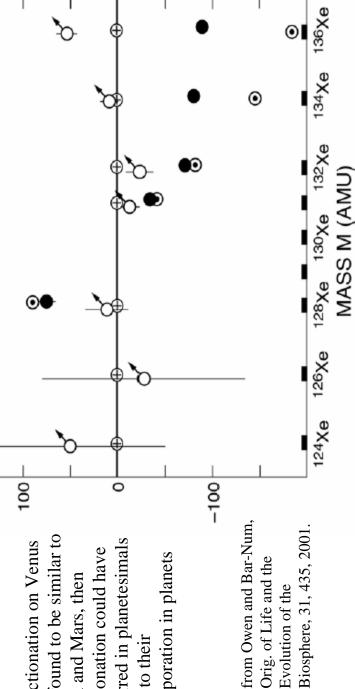
Solar Wind

Earth

Cl Chondrites

The fractionation in Venus is unknown.

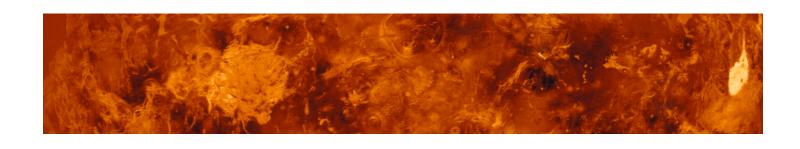
was found to be similar to occurred in planetesimals If fractionation on Venus fractionation could have incorporation in planets Earth and Mars, then prior to their



Evolution of the

Current status of noble gas measurements at Venus

Kr - no isotope information, great uncertainty in abundance Xe – no isotope information, upper limit on abundance



Present state of the art in Venus noble gas measurements

Noble gas	Previous	notes
abundance	measurements	
He	12 (+24,-8) ppm	12 (+24,-8) ppm extrapolated from meas. $> 130 \text{ km}$
Ne	7 ± 3 ppm	4 MS measurements
Ar	$70 \pm 25 \text{ ppm}$	3 MS and 2 GC measurements
	0.4 ± 0.14	Venera 11 and 12 reproduced
, A		measurements
2	0.2	PV Probe Hoffman analysis
	0.025	PV Probe Donahue analysis
Xe	0.12 upper limit	PV Probe Donahue analysis

accuracy

Target

<5-10%

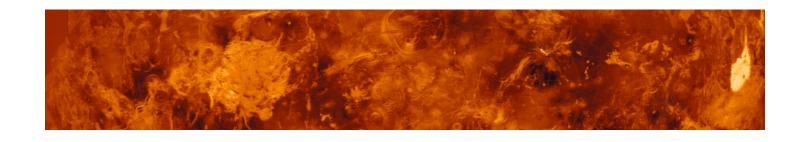
Target

measurements →
Kr and Xe
abundance and precision <1-2% distribution Key future isotopic

۱			
		-	Xe isotopes
•			Kr isotopes
5	Venera 11/12 MS	1.19 ± 0.07	W /W
•	PV Probe Donahue analysis	1.03 ± 0.04	40 A = 136 A =
_	Venera 11/12 MS	5.08 ± 0.05	
_	PV Probe Donahue analysis	5.56 ± 0.62	$^{36}\mathrm{Ar}/^{38}\mathrm{Ar}$
l ì			$^{20}\mathrm{Ne}/^{21}\mathrm{Ne}$
	Fotential interference from 40 Ar ⁺⁺ at 20 Da and CO ₂ ⁺⁺ at 22 Da	11.8 ± 0.7	$^{20}\mathrm{Ne}/^{22}\mathrm{Ne}$
	interference with HD		
	methane or H ₂ could give H ₃ ⁺		$^3{ m He}/^4{ m He}$
	3 He predicted at low ppb level –		
			ratio
		measurement	isotope
	notes	Previous	Noble gas

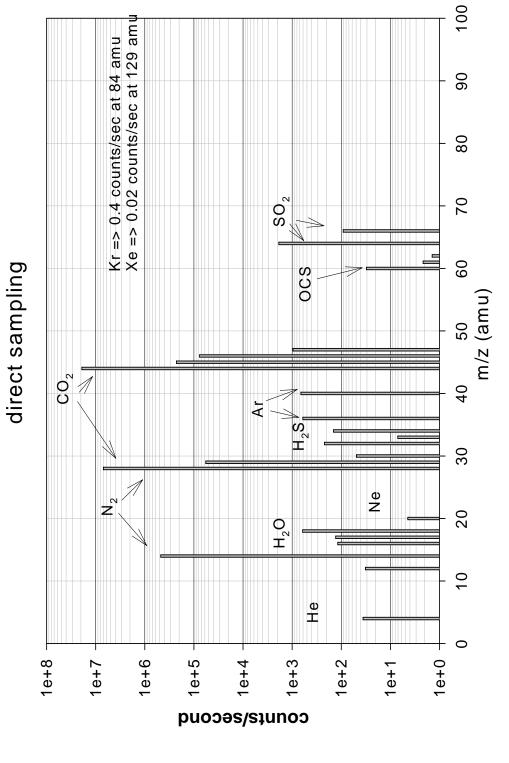
Approach for future noble gas measurements at Venus

Dedicated noble gas processing unit to optimize all noble gas Wide dynamic range mass spectrometer measurements including Xe and Kr

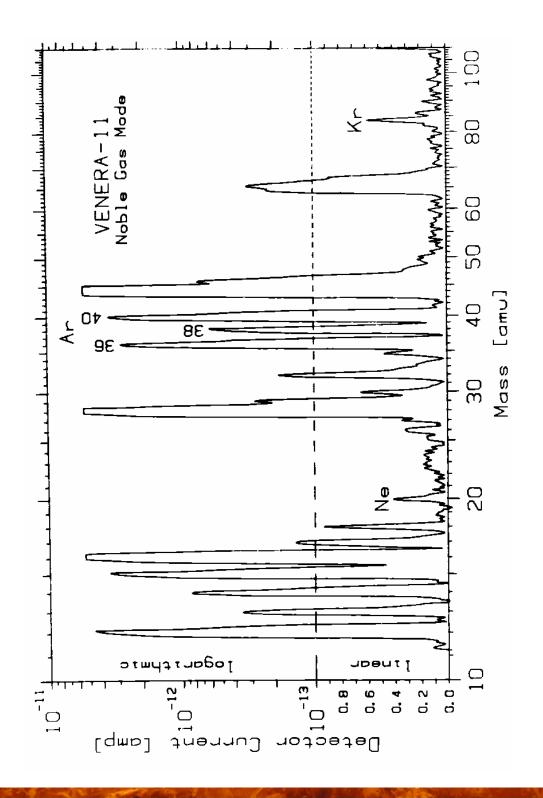


Predicted signal with direct sampling at Venus with no

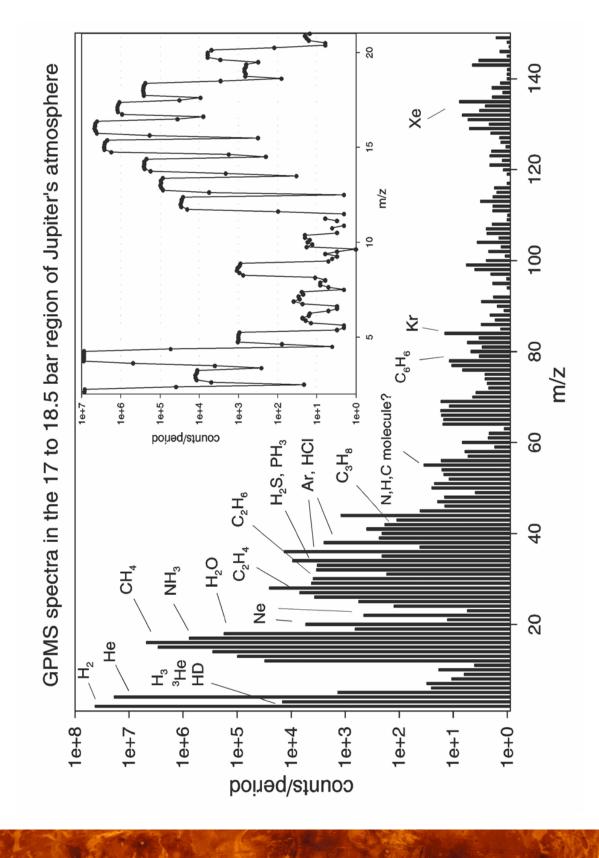




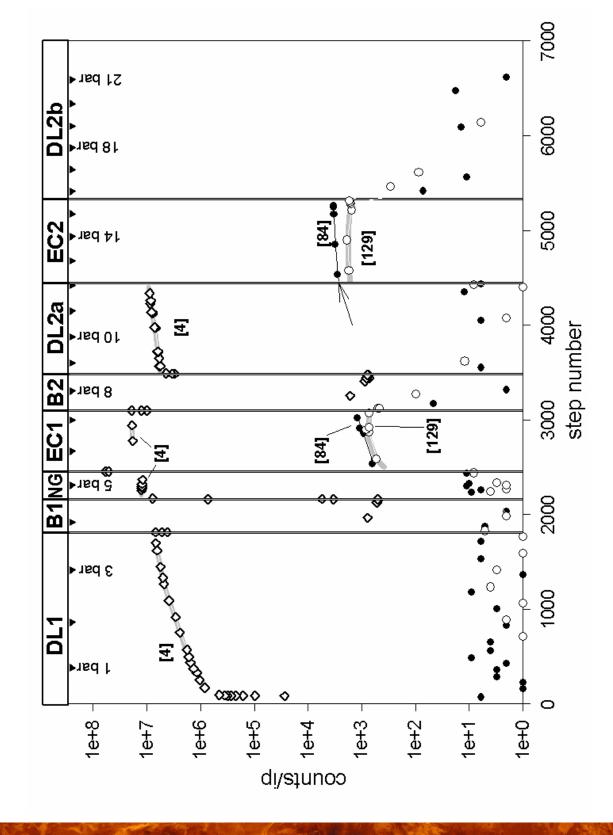




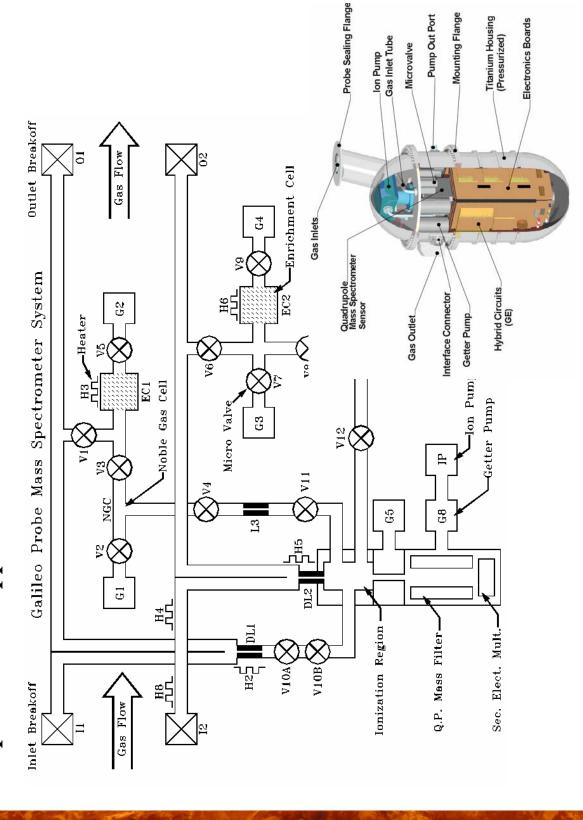
Dynamic range possible with small quadrupole mass spectrometer

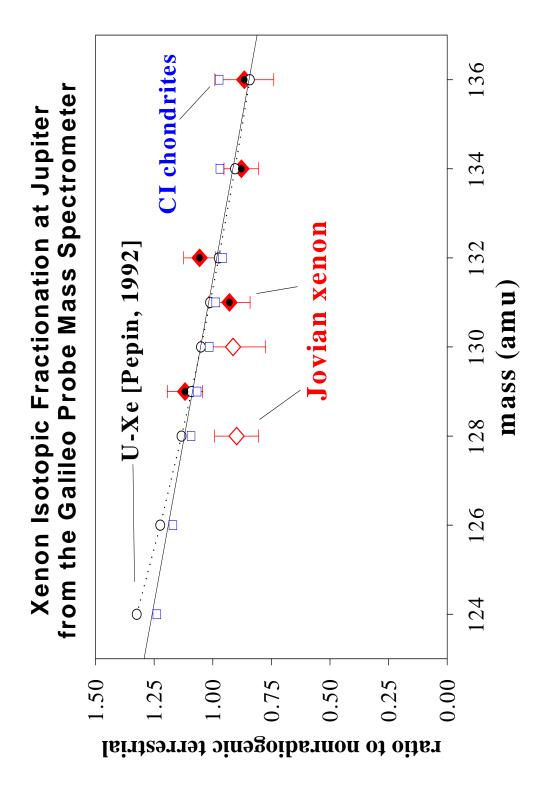


Galileo Probe use enrichment but NOT static MS



Enrichment techniques – the Galileo Probe Neutral Mass Spectrometer approach

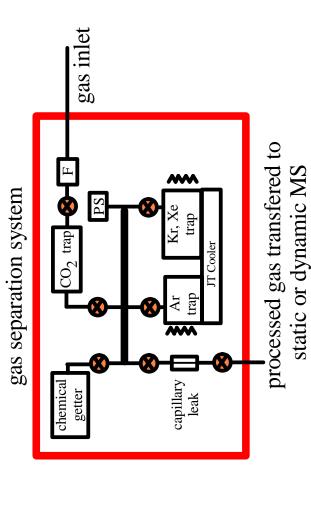






A proposed measurement protocol for Venus noble gas and ¹⁵N/¹⁴N measurement

- sample a volume of Venus atmospheric gas
- chemically remove CO₂ as gas is sampled
- (for example, CaO (s) + CO₂(g) \rightarrow CaCO₂ (s)
- chemically remove N₂ and other active gases with a getter • $(^{15}N^{14}N)/^{14}N_2$ with dynamic MS to obtain $^{15}N/^{14}N$
- cryogenically remove Kr and Xe (on high surface area trap)
 - 38 Ar/ 36 Ar and 36 Ar/ 40 Ar with static MS
- cryogenically remove Ar
- residual 20 Ne/ 22 Ne and ²¹Ne/²²Ne and ³He/⁴He with static MS
- release Kr and Xe
- all Kr and Xe isotopes with static MS

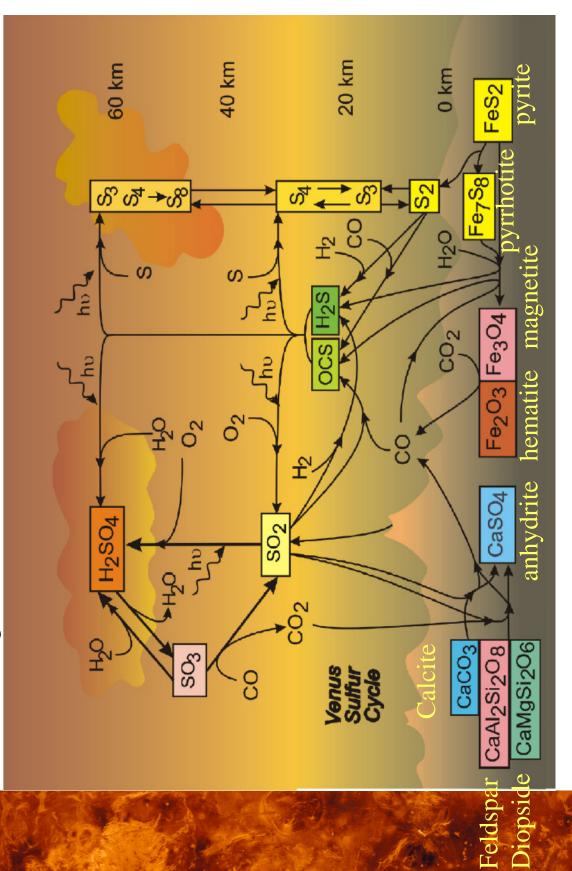


Motivation for trace gas measurements at Venus

surface enable cloud chemistry and atmosphere/surface Vertical profiles through the clouds and down to the interactions to be studied



S cycle - B. Fegley et al., in Venus II, U. AZ Press, 618 (1997) (following van Zahn & Prinn).



Gases and reactions expected to be important for cloud chemistry

SO₂, H₂O, SO₃, SO, OCS

$$SO_2 + \frac{1}{2}O_2 + H_2O + M \rightarrow H_2SO_4$$
 net reaction



Photolysis of $SO_2 \rightarrow SO + O$

Elemental sulfur

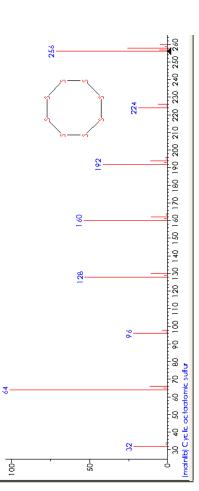
$$SO + SO \rightarrow SO_2 + S$$

$$S + S + M \rightarrow S_2 + M$$

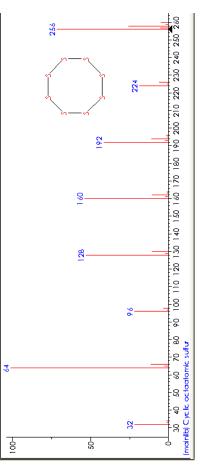
$$S_2 + S_2 + M \rightarrow S_4 + M$$

$$S_4 + S_4 + M \rightarrow S_8 + M$$

$$S_A + S_A + M \rightarrow S_8 + M$$







Reactions that may be important for surface/atmosphere interaction

Volcanoes likely source of SO,

Weathering of surface minerals may buffer atmospheric gases

$$CaCO_3(s) + SO_2(g) \rightarrow CaSO_4(s) + CO(g)$$

anhydrite Calcite

 $CaCO_{3}(s) + SiO_{2}(s) = CaSiO_{3}(s) + CO_{2}(g)$ (source of calcite - Fegley & Treiman, 1992) wollastonite quartz Calcite

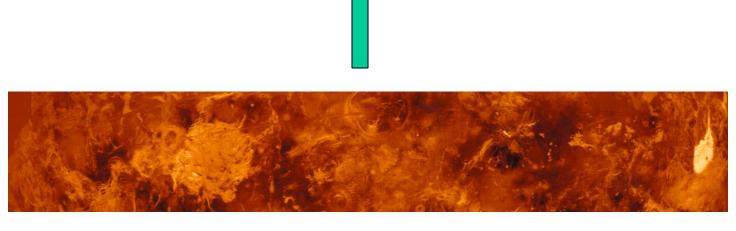
Trace species of interest that reflect the oxidation state near the surface

H₂S, SO₂, OCS, O₂, CO, H₂O

Oxidation state determines Fe mineralogy hematite $Fe_3O_4(s) + O_2 = Fe_2O_3(s)$ magnetite

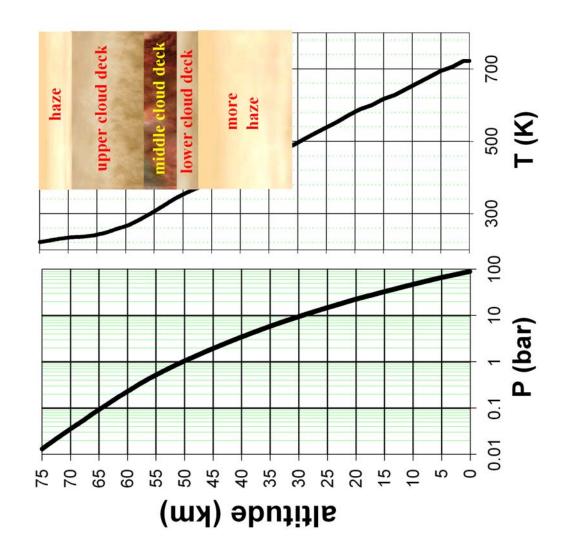
Past and future Venus mass spectrometer experiments





Sampling issues

- 4 orders of magnitude pressure differential on track from above clouds to surface
- trace species measured to parts per billion
- 9 orders of magnitude difference between atmospheric pressure at surface and ion source pressure in mass spectrometer
- 500 degree temperature gradient from atmosphere above clouds to surface
- cloud droplets and aerosols that can clog mass spectrometer inlet systems and mask real vertical variations due to their condensation on surfaces



Example Venus mass spectrometer experiments

Mission (Team,	Mass	Altitude	Inlet type	Outcome
Date)	Spectrome	(Pressure)		
	ter			
Venera 9 & 10	monopole	63-34 km	3 porous	instrument measured primarily
(Surkov, von		(130 mbar	sgnld	background signal throughout
Zahn, 1975)		to 6 bar		descent
PV-Large Probe	magnetic	62 km to	pinched Ta	50 km to 29 km inlet was blocked
(Hoffman, 1978)	sector	surface	tube (3 inlets)	and instrument measured outgassing
				from H ₂ SO ₄ droplets
Venera 11 & 12	Bennett RF	23 km to	1 m x 5 mm	possible inlet tube memory effects,
Lander (Grechnev,		surface	inlet pipe &	Ar isotopes in "static" mode, Kr
1978)			pulsed	detected but isotopes NOT resolved
			microvalve	
PV-Orbiter	Quadrupol	orbiter	source open	14 years of data →neutral scale
(Niemann,	e MS	(upper	to ambient	heights (CO ₊ , CO, N ₂ , O, N, and He)
Kasprzak, 1978-		atmosphere)		O escape (thermospheric
1992)				measurements gave no information
				on heavy noble gas isotopes)
PV-Multiprobe	Magnetic	entry to	open with	entry measurements (upper limit on
Bus (von Zahn,	Sector	0.01 mbar	differential P	36 Ar and 40 Ar), identified He
1978)				homopause at 137 km

Atmospheric sampling approach

- short inlet lines heated above ambient to vaporize condensates
- chemically inert materials in inlet
- adequate aerosol traps and baffles
- multiple inlet leaks
- redundant inlet lines

TEST TEST TEST TEST

